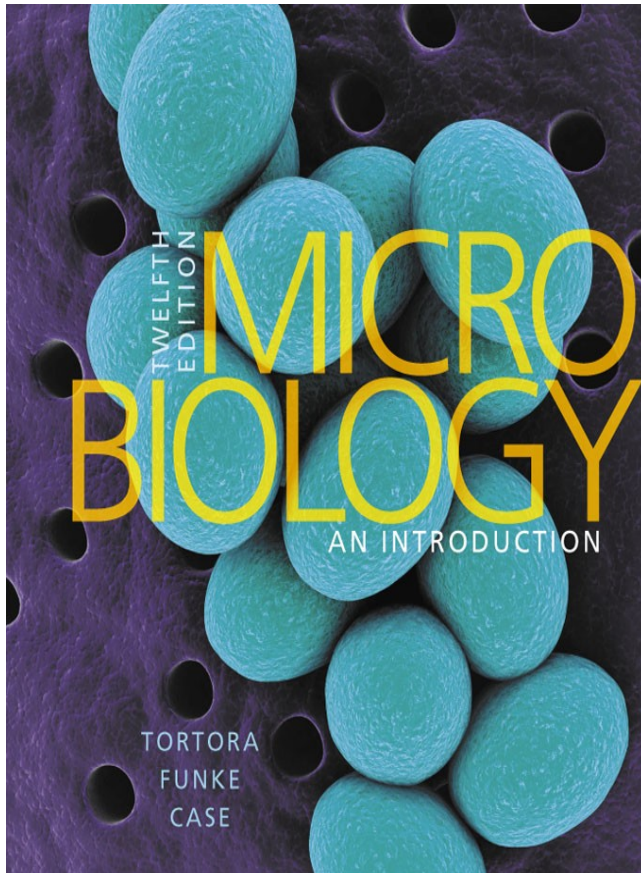


# Microbiology an Introduction

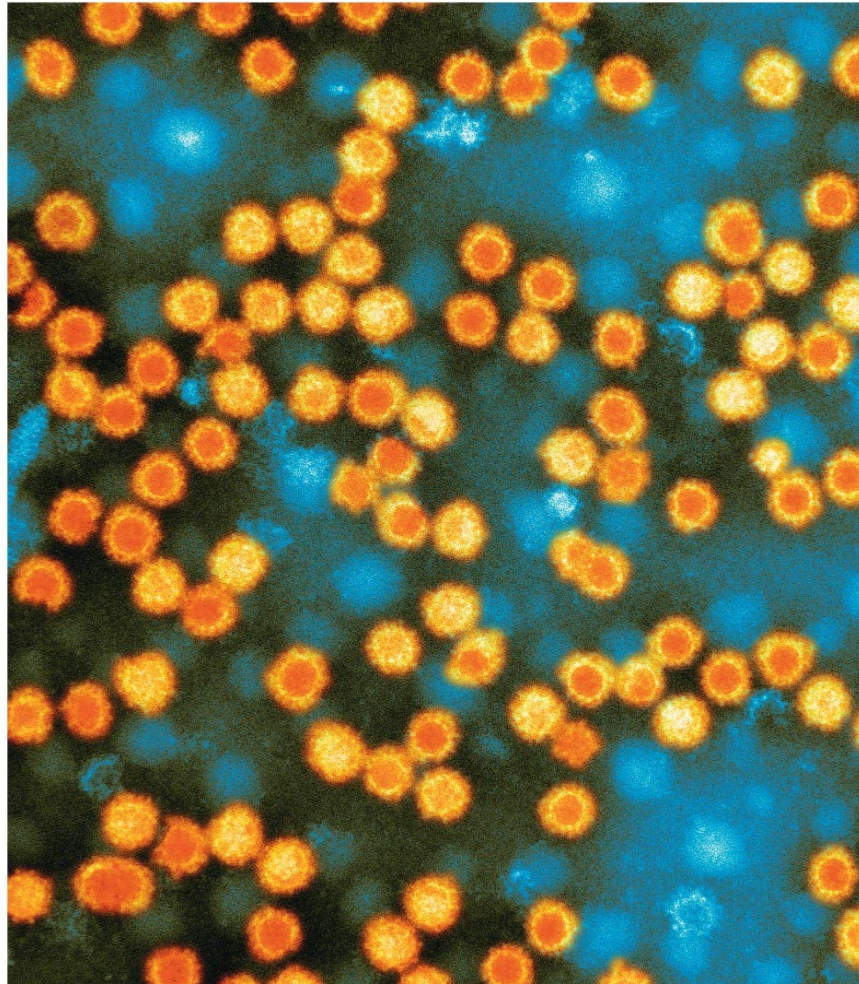
Twelfth Edition



## Chapter 7

### The Control of Microbial Growth

# Norovirus



# The Terminology of Microbial Control (1 of 4)

## Learning Objective

7-1 Define the following key terms related to microbial control: **sterilization, disinfection, antisepsis, degerming, sanitization, biocide, germicide, bacteriostasis, and asepsis.**

# The Terminology of Microbial Control

(2 of 4)

- **Sepsis** refers to bacterial contamination
- **Asepsis** is the absence of significant contamination
  - Aseptic surgery techniques prevent the microbial contamination of wounds

# The Terminology of Microbial Control

(3 of 4)

- **Sterilization:** removing and destroying all microbial life
- **Commercial sterilization:** killing **C. botulinum** endospores from canned goods
- **Disinfection:** destroying harmful microorganisms
- **Antisepsis:** destroying harmful microorganisms from living tissue

# The Terminology of Microbial Control

(4 of 4)

- **Degerming:** the mechanical removal of microbes from a limited area
- **Sanitization:** lowering microbial counts on eating utensils to safe levels
- **Biocide (germicide):** treatments that kill microbes
- **Bacteriostasis:** inhibiting, not killing, microbes

# Check Your Understanding-1

## Check Your Understanding

- ✓ The usual definition of **sterilization** is the removal or destruction of all forms of microbial life; how could there be practical exceptions to this simple definition?

7-1

# The Rate of Microbial Death (1 of 3)

## Learning Objective

7-2 Describe the patterns of microbial death caused by treatments with microbial control agents.



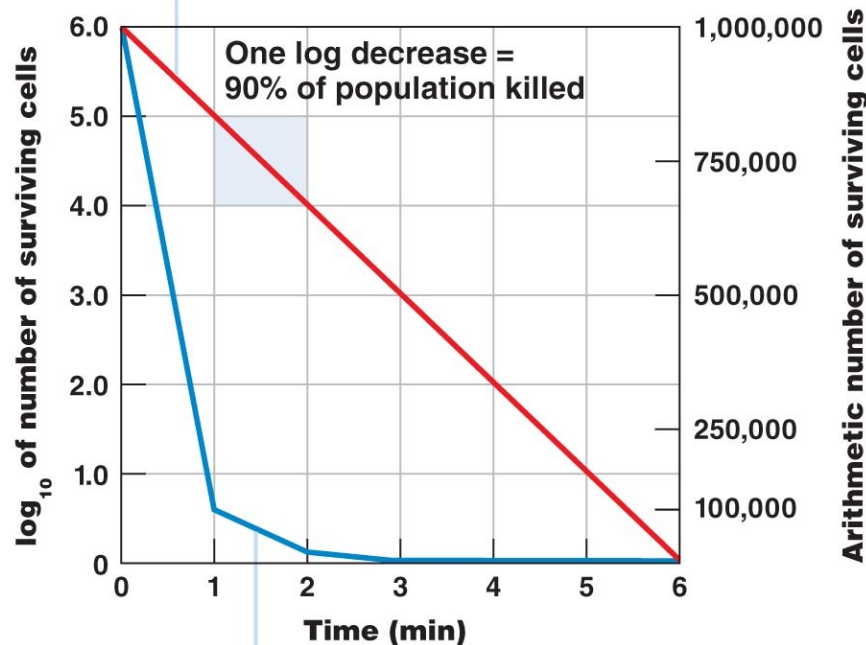
# The Rate of Microbial Death (2 of 3)

**Table 7.2 Microbial Exponential Death Rate: An**

| <b>Time (min)</b> | <b>Deaths per Minute</b> | <b>Number of Survivors</b> |
|-------------------|--------------------------|----------------------------|
| 0                 | 0                        | 1,000,000                  |
| 1                 | 900,000                  | 100,000                    |
| 2                 | 90,000                   | 10,000                     |
| 3                 | 9000                     | 1000                       |
| 4                 | 900                      | 100                        |
| 5                 | 90                       | 10                         |
| 6                 | 9                        | 1                          |

# Figure 7.1a Understanding the Microbial Death Curve

Plotting the typical microbial death curve **logarithmically** (**red line**) results in a straight line.

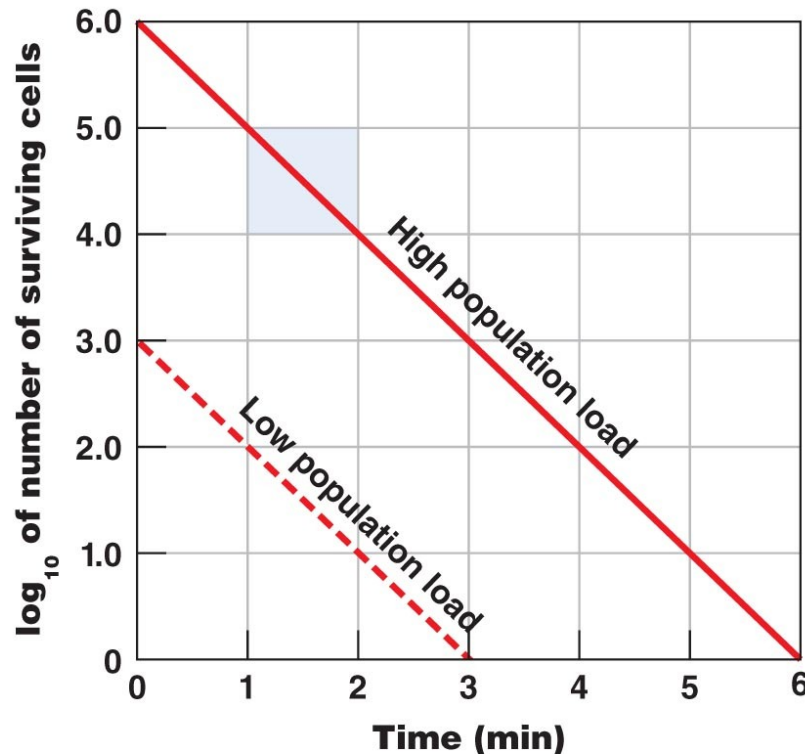


**(a)** Plotting the typical microbial death curve **arithmetically** (**blue line**) is impractical: at 3 minutes the population of 1000 cells would only be a hundredth of the graphed distance between 100,000 and the baseline.

# The Rate of Microbial Death (3 of 3)

- Effectiveness of treatment depends on:
  - Number of microbes
  - Environment (organic matter, temperature, biofilms)
  - Time of exposure
  - Microbial characteristics

# Figure 7.1b Understanding the Microbial Death Curve



**(b)** Logarithmic plotting (**red**) reveals that if the rate of killing is the same, it will take longer to kill all members of a larger population than a smaller one, whether using heat or chemical treatments.

# Check Your Understanding-2

## Check Your Understanding

- ✓ How is it possible that a solution containing a million bacteria would take longer to sterilize than one containing a half-million bacteria?  
7-2

# **Actions of Microbial Control Agents** (1 of 2)

## **Learning Objective**

7-3 Describe the effects of microbial control agents on cellular structures.

# Actions of Microbial Control Agents (2 of 2)

- Alteration of membrane permeability
- Damage to proteins (enzymes)
- Damage to nucleic acids

# Check Your Understanding-3

## Check Your Understanding

- ✓ Would a chemical microbial control agent that affects plasma membranes affect humans?  
7-3



# Physical Methods of Microbial Control (1 of 2)

## Learning Objectives

7-4 Compare the effectiveness of moist heat (boiling, autoclaving, pasteurization) and dry heat.

7-5 Describe how filtration, low temperatures, high pressure, desiccation, and osmotic pressure suppress microbial growth.

7-6 Explain how radiation kills cells.

# Heat (1 of 3)

- Heat denatures enzymes
- **Thermal death point (TDP):** lowest temperature at which all cells in a liquid culture are killed in 10 min
- **Thermal death time (TDT):** minimal time for all bacteria in a liquid culture to be killed at a particular temperature

# Heat (2 of 3)

- **Decimal reduction time (DRT)**
  - Minutes to kill 90% of a population at a given temperature

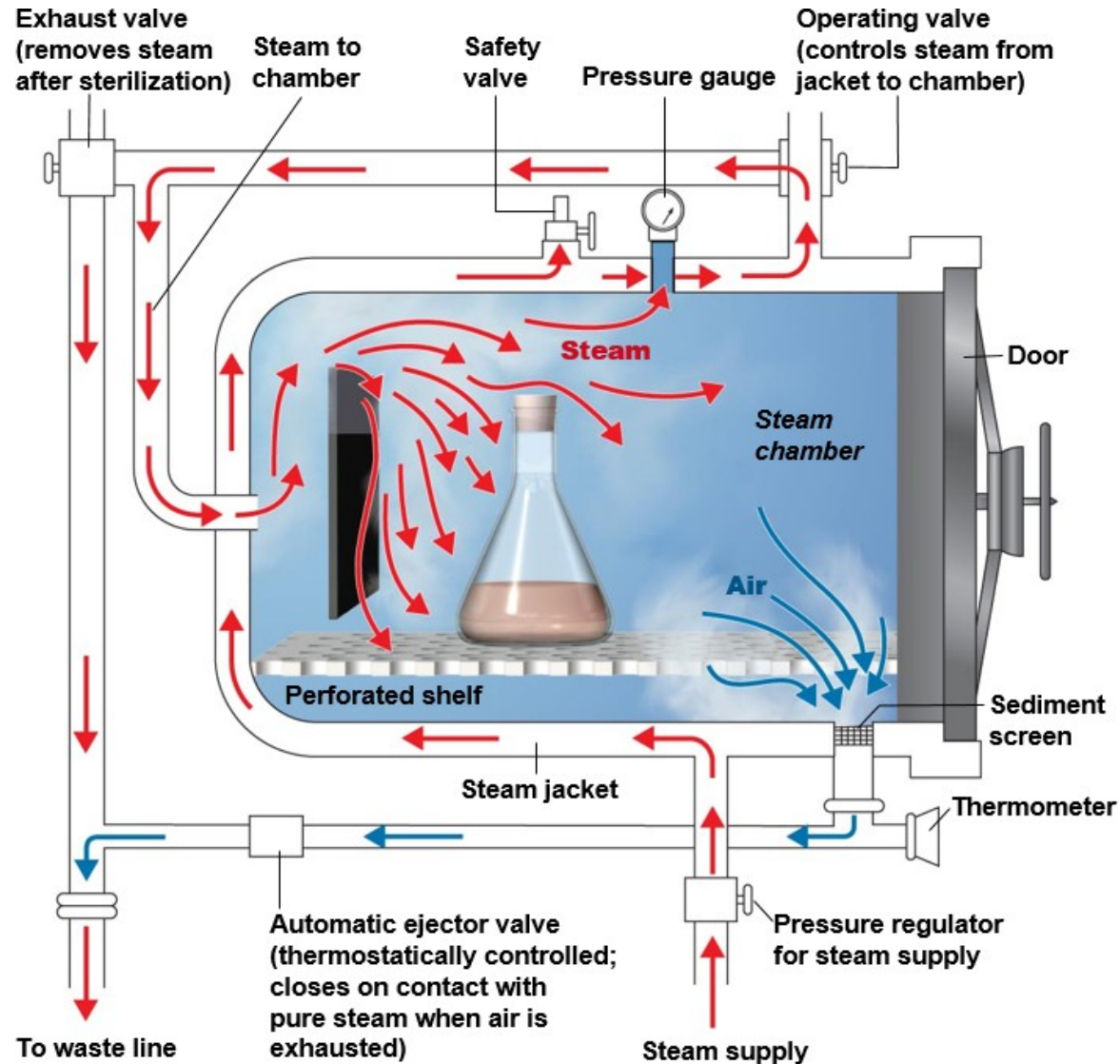
# Moist Heat Sterilization (1 of 4)

- Moist heat denatures proteins
- Boiling
- Free-flowing steam

# Moist Heat Sterilization (2 of 4)

- **Autoclave:** steam under pressure
- 121 C at 15 psi for 15 min
- Kills all organisms and endospores
- Steam must contact the item's surface

# Figure 7.2 An Autoclave



# Moist Heat Sterilization (3 of 4)

- Large containers require longer sterilization times
- Test strips are used to indicate sterility

# Moist Heat Sterilization (4 of 4)

**Table 7.4 The Effect of Container Size on Autoclave Sterilization Times for Liquid Solutions\***

| <b>Pressure (psi in Excess<br/>of Atmospheric<br/>Pressure)</b> | <b>Temperature (°C)</b> |
|---|-------------------------|
| 0   | 100                     |
| 5   | 110                     |
| 10  | 116                     |
| 15  | 121                     |
| 20  | 126                     |
| 30  | 135                     |

\*At higher altitudes, the atmospheric pressure is less, a phenomenon that must be taken into account in operating an autoclave. For example, to reach sterilizing temperatures (121°C) in Denver, Colorado, whose altitude is 5280 feet (1600 meters), the pressure shown on the autoclave gauge would need to be higher than the 15 psi shown in the table.



# Figure 7.3 Examples of Sterilization Indicators



# Heat (3 of 3)

- **Pasteurization** reduces spoilage organisms and pathogens
- Equivalent treatments
  - 63°C for 30 min
  - **High-temperature short-time (HTST):** 72°C for 15 sec
  - **Ultra-high-temperature (UHT):** 140°C for 4 sec
- **Thermoduric** organisms survive

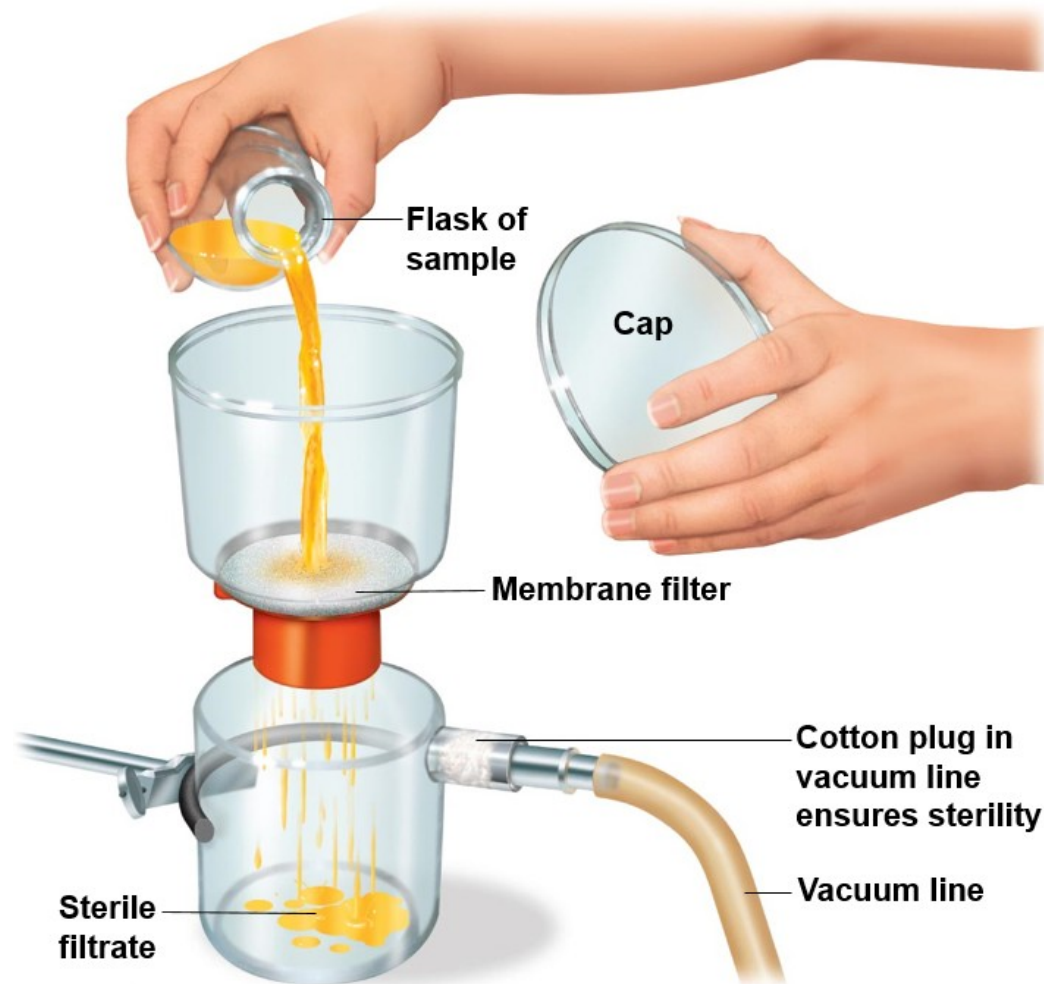
# Dry Heat Sterilization

- Kills by oxidation
  - **Flaming**
  - Incineration
  - **Hot-air sterilization**

# Filtration

- Passage of substance through a screenlike material
- Used for heat-sensitive materials
- **High-efficiency particulate air (HEPA) filters** remove microbes  $> 0.3 \mu\text{m}$
- **Membrane filters** remove microbes  $> 0.22 \mu\text{m}$

# Figure 7.4 Filter Sterilization with a Disposable, Presterilized Plastic Unit



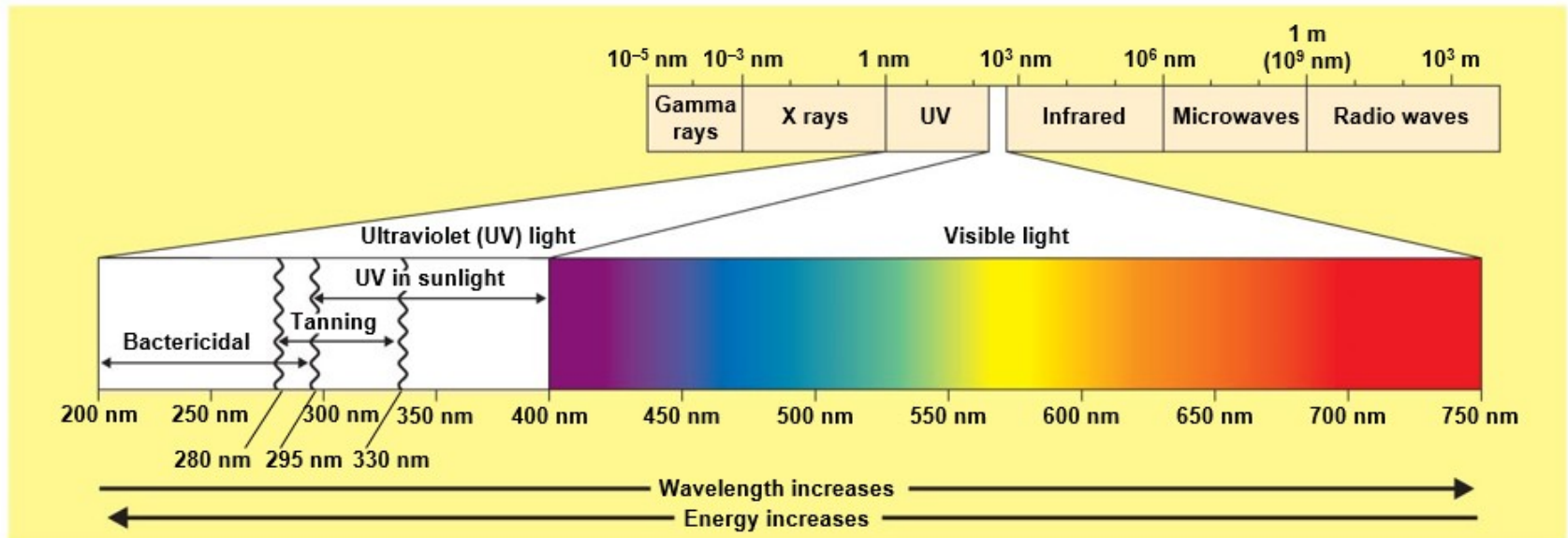
# Physical Methods of Microbial Control (2 of 2)

- Low temperature has a bacteriostatic effect
  - Refrigeration
  - Deep-freezing
  - Lyophilization (freeze drying)
- High pressure denatures proteins
- **Desiccation:** absence of water prevents metabolism
- Osmotic pressure uses salts and sugars to create hypertonic environment; causes plasmolysis

# Radiation (1 of 2)

- **Ionizing radiation** (X rays, gamma rays, electron beams)
  - Ionizes water to create reactive hydroxyl radicals
  - Damages DNA by causing lethal mutations
- **Nonionizing radiation** (UV, 260 nm)
  - Damages DNA by creating thymine dimers
- **Microwaves** kill by heat; not especially antimicrobial

# Figure 7.5 The Radiant Energy Spectrum





# Check Your Understanding-4

## Check Your Understanding

- ✓ How is microbial growth in canned foods prevented?  
7-4
- ✓ Why would a can of pork take longer to sterilize at a given temperature than a can of soup that also contained pieces of pork?  
7-5
- ✓ What is the connection between the killing effect of radiation and hydroxyl radical forms of oxygen?  
7-6

# Chemical Methods of Microbial Control (1 of 2)

## Learning Objectives

7-7 List the factors related to effective disinfection.

7-8 Interpret the results of use-dilution tests and the disk diffusion method.

# Principles of Effective Disinfection

- Concentration of disinfectant
- Organic matter
- pH
- Time

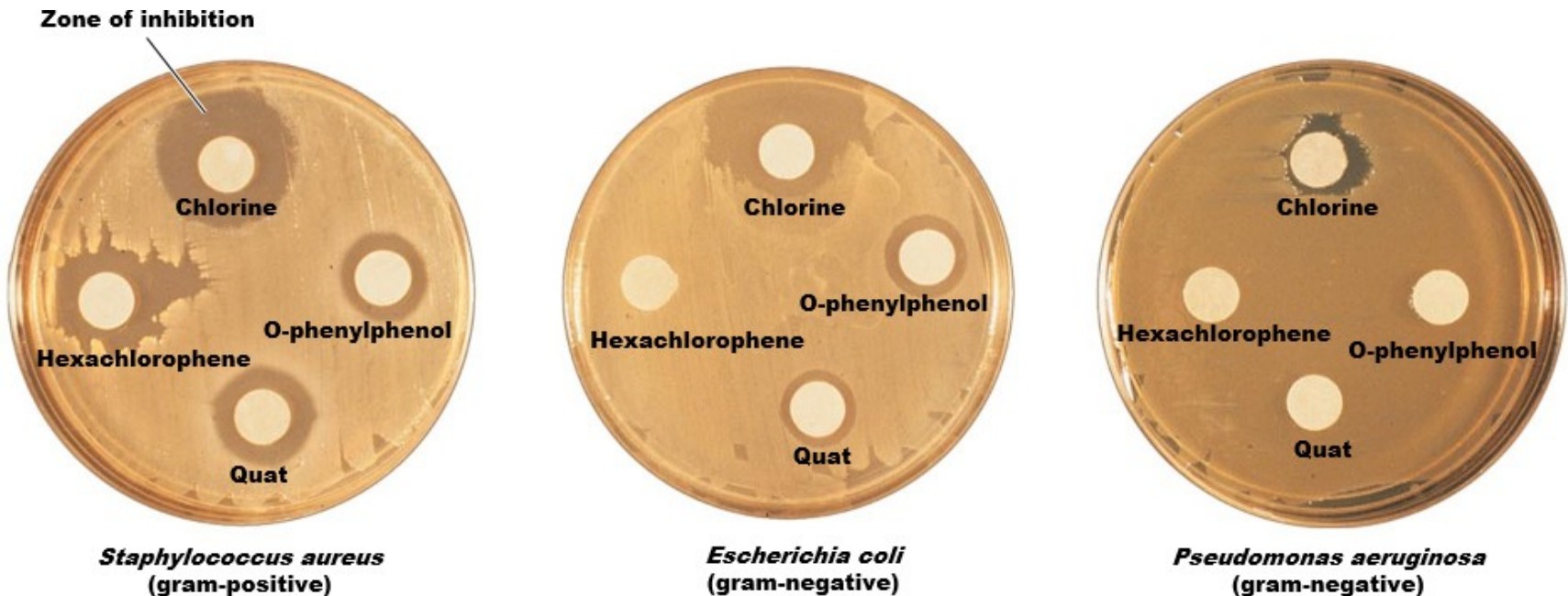
# Use-Dilution Tests

- Metal cylinders are dipped in test bacteria and dried
- Cylinders are placed in disinfectant for 10 min at 20°C
- Cylinders are transferred to culture media to determine whether the bacteria survived treatment

# The Disk-Diffusion Method

- Evaluates efficacy of chemical agents
- Filter paper disks are soaked in a chemical and placed on a culture
- Look for zone of inhibition around disks

# Figure 7.6 Evaluation of Disinfectants by the Disk-Diffusion Method



# Check Your Understanding-5

## Check Your Understanding

- ✓ If you wanted to disinfect a surface contaminated by vomit and a surface contaminated by a sneeze, why would your choice of disinfectant make a difference?  
7-7
- ✓ Which is more likely to be used in a medical clinic laboratory, a use-dilution test or a disk-diffusion test?  
7-8

# Chemical Methods of Microbial Control (2 of 2)

## Learning Objectives

7-9 Identify the methods of action and preferred uses of chemical disinfectants.

7-10 Differentiate halogens used as antiseptics from halogens used as disinfectants.

7-11 Identify the appropriate uses for surface-active agents.

7-12 List the advantages of glutaraldehyde over other chemical disinfectants.

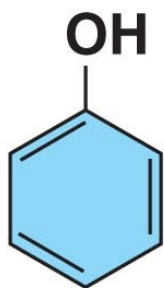
7-13 Identify chemical sterilizers.



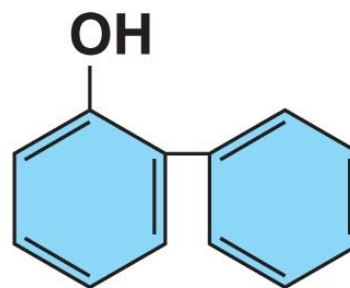
# Phenol and Phenolics

- Injure lipids of plasma membranes, causing leakage

# Figure 7.7a-b The Structure of Phenolics and Bisphenols



**(a) Phenol**

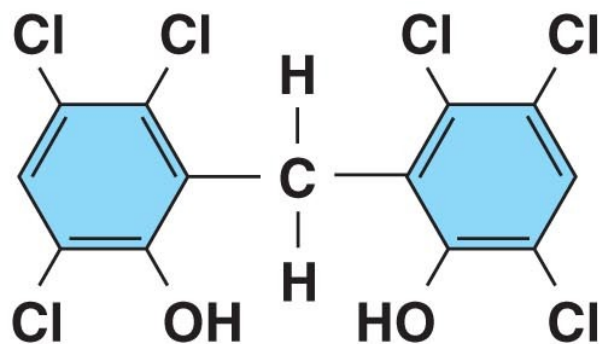


**(b) O-phenylphenol**

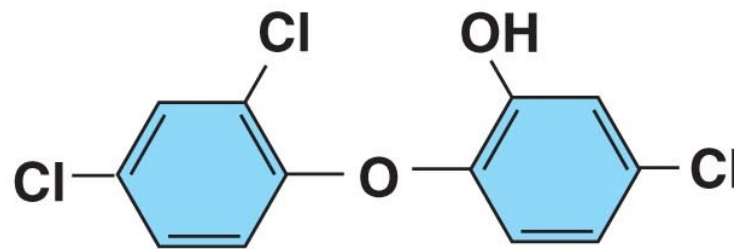
# Bisphenols

- Contain two phenol groups connected by a bridge
- Hexachlorophene and triclosan
- Disrupt plasma membranes

# Figure 7.7c-d The Structure of Phenolics and Bisphenols



**(c) Hexachlorophene (a bisphenol)**



**(d) Triclosan (a bisphenol)**

# Biguanides

- Chlorhexidine
- Used in surgical hand scrubs
- Disrupt plasma membranes

# Halogens

- Iodine
  - **Tincture:** solution in aqueous alcohol
  - **Iodophor:** combined with organic molecules
  - Impairs protein synthesis and alters membranes
- Chlorine
  - Oxidizing agents; shut down cellular enzyme systems
  - Bleach: hypochlorous acid (HOCl)
  - Chloramine: chlorine + ammonia

# Alcohols (1 of 2)

- Denature proteins and dissolves lipids
- No effect on endospores and nonenveloped viruses
- Ethanol and isopropanol
  - Require water

# Alcohols (2 of 2)

**Table 7.6 Biocidal Action of Various Concentrations of Ethanol in Aqueous Solution against Streptococcus pyogenes**

| Concentration of Ethanol (%) | 10 | 20 | 30 | 40 | 50 |
|------------------------------|----|----|----|----|----|
| 100                          | G  | G  | G  | G  | G  |
| 95                           | NG | NG | NG | NG | NG |
| 90                           | NG | NG | NG | NG | NG |
| 80                           | NG | NG | NG | NG | NG |
| 70                           | NG | NG | NG | NG | NG |
| 60                           | NG | NG | NG | NG | NG |
| 50                           | G  | G  | NG | NG | NG |
| 40                           | G  | G  | G  | G  | G  |

**Note:**

G = growth

NG = no growth



# Heavy Metals and Their Compounds

- **Oligodynamic action**—very small amounts exert antimicrobial activity
- Denature proteins
- Ag, Hg, Cu, Zn
  - Silver nitrate is used to prevent ophthalmia neonatorum
  - Mercuric chloride prevents mildew in paint
  - Copper sulfate is an algicide
  - Zinc chloride is found in mouthwash

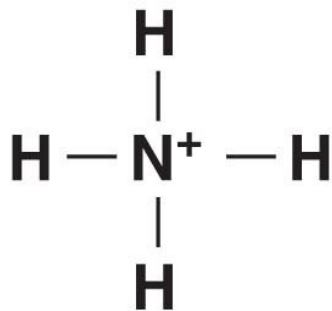
# Figure 7.8 Oligodynamic Action of Heavy Metals



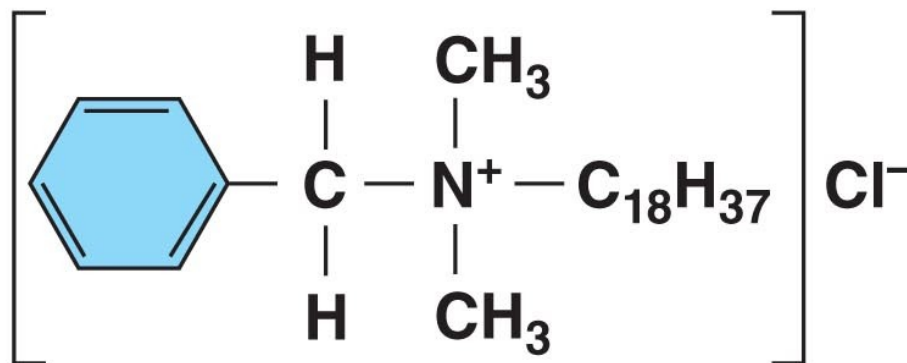
# Surface-Active Agents

|  |   |
|--|---|
| Soap   | Degerming;<br>emulsification  |
| Acid-anionic sanitizers                              | Anions react with<br>plasma membrane  |
| <b>Quaternary<br/>ammonium<br/>compounds (quats)</b> | Cations are<br>bactericidal, denature<br>proteins, disrupt<br>plasma membrane |

# Figure 7.9 The Ammonium Ion and a Quaternary Ammonium Compound, Benzalkonium Chloride (Zephiran)

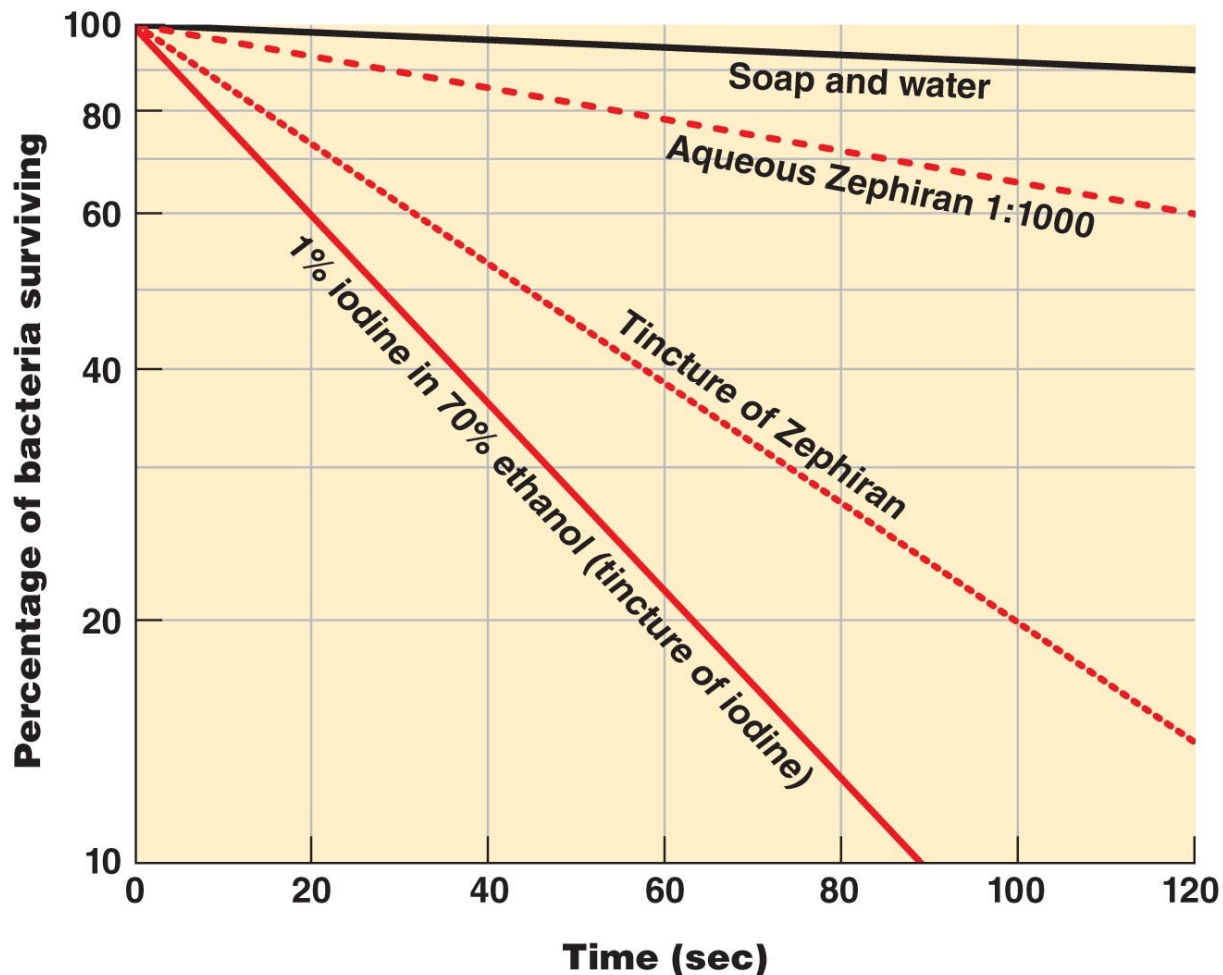


Ammonium ion



Benzalkonium chloride

# Figure 7.10 A Comparison of the Effectiveness of Various Antiseptics

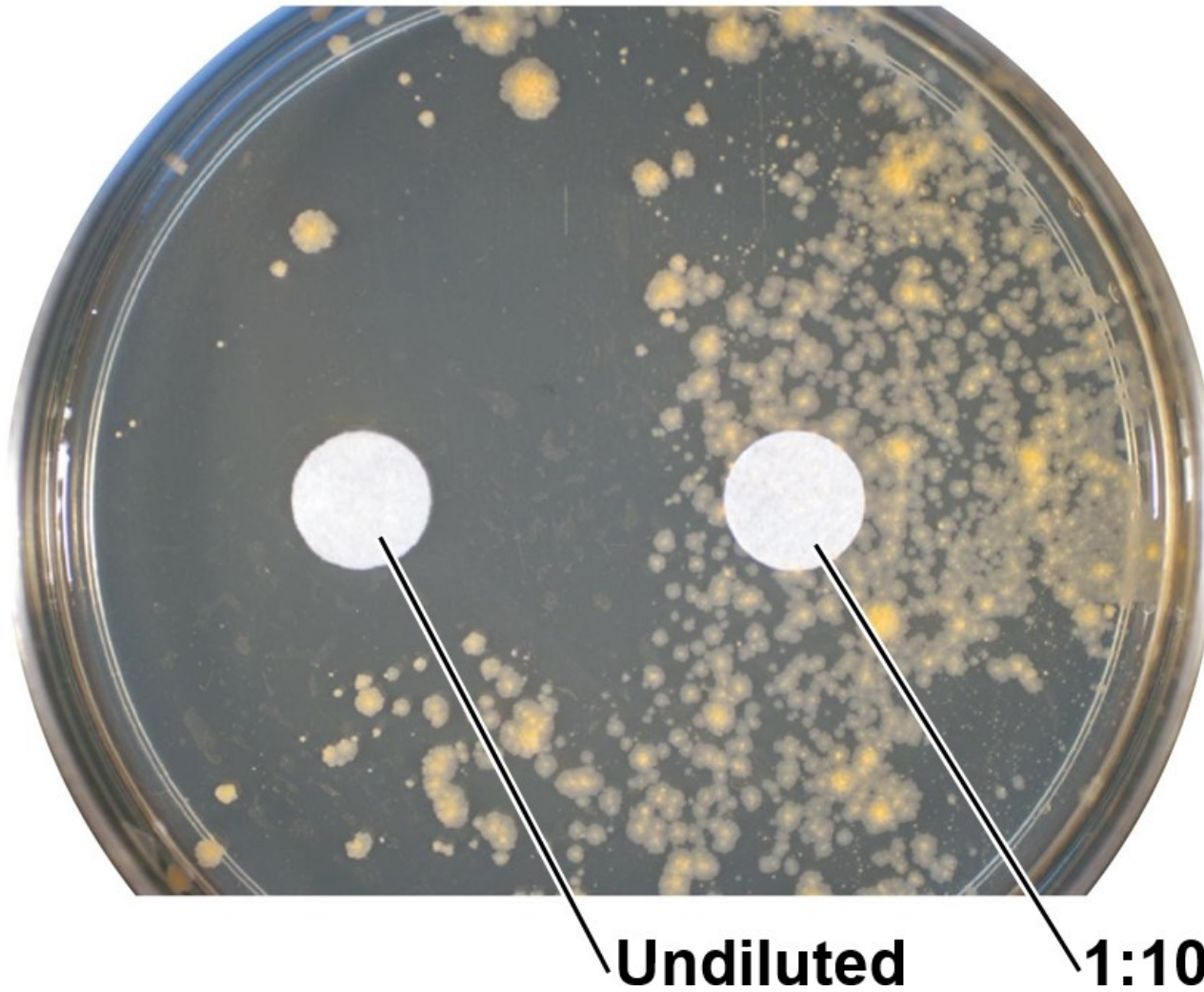


# **Clinical Focus: Infection Following Anesthesia Injection**

- Which preparation is more effective?

# Clinical Focus

## 7.1



# Chemical Food Preservatives

- Sulfur dioxide prevents wine spoilage
- Organic acids
  - Inhibit metabolism
  - Sorbic acid, benzoic acid, and calcium propionate prevent molds in acidic foods
- Nitrites and nitrates prevent endospore germination



# Antibiotics

- Bacteriocins—proteins produced by one bacterium that inhibits another
- Nisin and natamycin prevent spoilage of cheese

# Aldehydes

- Inactivate proteins by cross-linking with functional groups, -NH<sub>2</sub>, -OH, -COOH, -SH
- Used for preserving specimens and in medical equipment
  - Formaldehyde and ortho-phthalaldehyde
  - Glutaraldehyde is one of the few liquid chemical sterilizing agents

# Chemical Sterilization

- Gaseous sterilants cause alkylation—replacing hydrogen atoms of a chemical group with a free radical
- Cross-links nucleic acids and proteins
- Used for heat-sensitive material
  - Ethylene oxide

# Plasma

- Fourth state of matter, consisting of electrically excited gas
- Free radicals destroy microbes
- Used for tubular instruments

# Supercritical Fluids

- CO<sub>2</sub> with gaseous and liquid properties
- Used for medical implants

# Peroxygens and Other Forms of Oxygen

- Oxidizing agents
- Used for contaminated surfaces and food packaging
  - $O_3$ ,  $H_2O_2$ , and peracetic acid

# Check Your Understanding-6

## Check Your Understanding

- ✓ Why is alcohol effective against some viruses and not others?  
7-9
- ✓ Is Betadine an antiseptic or a disinfectant when it is used on skin?  
7-10
- ✓ What characteristics make surface-active agents attractive to the dairy industry?  
7-11
- ✓ What chemical disinfectants can be considered sporicides?  
7-12



What chemicals are used to sterilize?

7-13

# Microbial Characteristics and Microbial Control (1 of 2)

## Learning Objective

7-14 Explain how the type of microbe affects the control of microbial growth.



# Microbial Characteristics and Microbial Control (2 of 2)

**Table 7.7 Effectiveness of Chemical Antimicrobials against Endospores and Mycobacteria**

| <b>Chemical Agent</b> | <b>Effect against Endospores</b> | <b>Effect against Mycobacteria</b> |
|-----------------------|----------------------------------|------------------------------------|
| Glutaraldehyde        | Fair                             | Good                               |
| Chlorines             | Fair                             | Fair                               |
| Alcohols              | Poor                             | Good                               |
| Iodine                | Poor                             | Good                               |
| Phenolics             | Poor                             | Good                               |
| Chlorhexidine         | None                             | Fair                               |
| Bisphenols            | None                             | None                               |
| Quats                 | None                             | None                               |
| Silver                | None                             | None                               |

# Check Your Understanding-7

## Check Your Understanding

- ✓ The presence or absence of endospores has an obvious effect on microbial control, but why are gram-negative bacteria more resistant to chemical biocides than gram-positive bacteria?  
7-14